

Synthesis and Interconversion of 5-Phenyl-1,3,2,4,6-dithiatriazine Derivatives; Crystal and Molecular Structure of the Bicyclic Molecule PhCN₅S₃

René T. Boéré,^a A. Wallace Cordes,^b and Richard T. Oakley^{a*}

^a Guelph Waterloo Centre for Graduate Work in Chemistry, Guelph Campus, Department of Chemistry and Biochemistry, University of Guelph, Guelph, Ontario N1G 2W1, Canada

^b Department of Chemistry, University of Arkansas, Fayetteville, Arkansas 72701, U.S.A.

Novel synthetic routes to heterocycles based on the 1,3,2,4,6-dithiatriazine framework are described; the X-ray crystal structure determination of the bicyclic derivative PhCN₅S₃ is reported.

Considerable interest has been demonstrated in the chemistry of heterocyclic thiazyl compounds, *i.e.* compounds that incorporate one or more unsaturated carbon atoms into an otherwise binary SN framework.¹⁻⁸ Most of the preparative routes to molecules of this type involve the reactions of nitriles and amidines (or their salts) with S₃N₃Cl₃, S₃N₂Cl₂, or SCl₂.²⁻⁸ However, these methods generally lead to a mixture of products, the composition of which is difficult to anticipate. Following our recent study² of the reaction of benzamidine with S₃N₃Cl₃ we have examined the use of *N,N,N'*-tris(trimethylsilyl)benzamidine PhC(NSiMe₃)N(SiMe₃)₂⁹ as a milder, more specific reagent. We report herein that the reaction of this compound with S₃N₃Cl₃ provides a simple, efficient, and rational synthesis of the bicyclic molecule PhCN₅S₃ (**1**).

Compound (**1**) was prepared by the slow addition of a solution of PhC(NSiMe₃)N(SiMe₃)₂ (6.89 g, 20.5 mmol) in 100 ml CH₂Cl₂ to a stirred solution of S₃N₃Cl₃ (5.00 g, 20.5 mmol) in 100 ml CH₂Cl₂ at 0 °C. Subsequent removal of the solvent and recrystallization of the residue from hot acetonitrile afforded golden air-stable plates of (**1**) {3.20 g, 12.5 mmol, 61%, decomp. 136 °C; † *m/z* [electron impact (E.I.), 70 eV] 195 (PhCN₃S₂⁺, 23%), 181 (PhCN₂S₂⁺, 75), 149 (PhCN₂S⁺, 29), 135 (PhCNS⁺, 30), 103 (PhCN⁺, 100)}.

The crystal and molecular structures of (**1**) have been determined by X-ray crystallography. ‡ An ORTEP drawing of the molecule and pertinent structural information are provided in Figure 1. The molecule can be described in terms of a 1,3,2,4,6-dithiatriazine skeleton loosely bridged by a NSN fragment across the 1,3-sulphur atoms. As such the molecule has many structural features in common with the phosphorus containing molecule F₂PS₃N₅.¹⁰ In particular we note that the bonds to the NSN bridge in (**1**) [mean of S(1)-N(3) and S(2)-N(2) = 1.728(2) Å] are even longer than the corresponding linkages in F₂PS₃N₅ [1.692(8) Å].

Oxidation of (**1**) (4.00 g, 15.7 mmol) in 100 ml CCl₄ with chlorine yields (in addition to small amounts of

‡ *Crystal data*: C₇H₅N₅S₃, *M* = 255.3, monoclinic, space group *P*2₁/*n*, *a* = 5.958(1), *b* = 22.955(2), *c* = 7.427(1) Å, β = 106.25°, *Z* = 4, *U* = 975.2 Å³, and *D_c* = 1.740 g cm⁻³. Data were collected on an Enraf-Nonius CAD-4 automated diffractometer with graphite monochromated Mo-*K*_α radiation (λ = 0.71073 Å) using ω-2θ scans (θ_{max.} of 50°), and were corrected for absorption. The structure was solved by direct methods and refined by full-matrix least-squares to an *R*-value of 0.030 for 1364 independent reflections (*I* > 3σ_{*i*}) and 156 parameters. All hydrogen atom positions were located from Fourier difference maps and were refined isotropically. The atomic coordinates are available on request from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW. Any request should be accompanied by the full literature citation for this communication.

† The elemental compositions of (**1**), (**2**), (**3**), and (**4**) have been confirmed by chemical analysis.

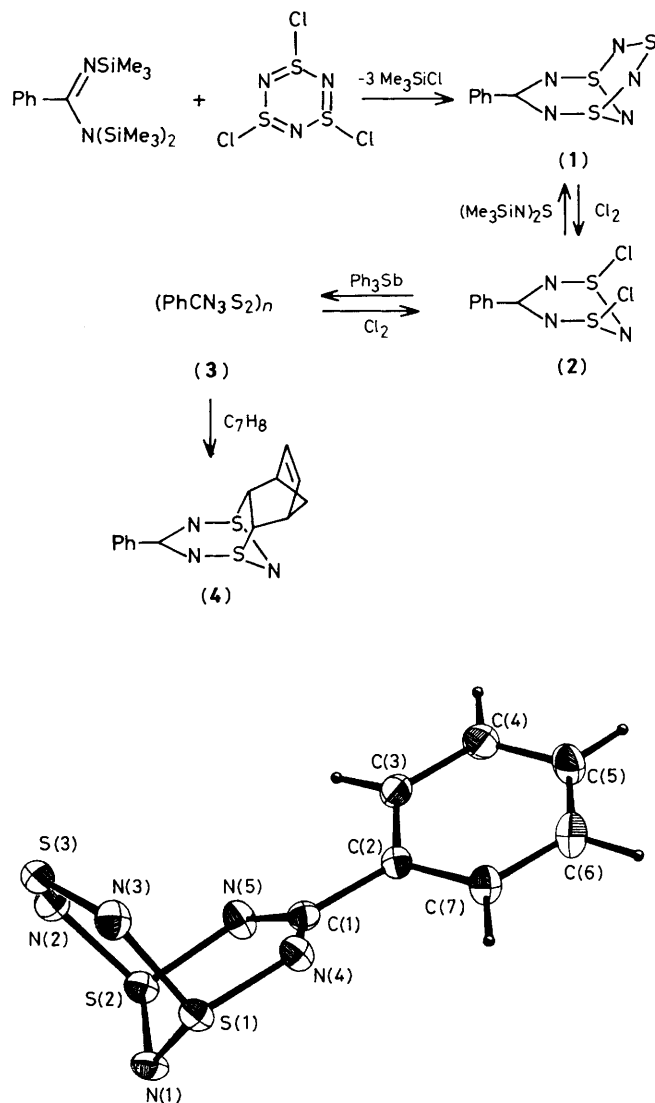


Figure 1. ORTEP drawing (30% probability ellipsoids) of PhCN_3S_3 (1). Selected bond lengths (in Å) and angles (in °) are: S(1)–N(1) 1.630(2), S(1)–N(3) 1.728(2), S(1)–N(4) 1.622(2), S(2)–N(1) 1.630(2), S(2)–N(2) 1.728(2), S(2)–N(5) 1.621(2), S(3)–N(2) 1.547(2), S(3)–N(3) 1.547(2), N(4)–C(1) 1.337(3), N(5)–C(1) 1.333(3), C(1)–C(2) 1.486(3); N(1)–S(1)–N(3) 106.1(1), N(1)–S(1)–N(4) 109.5(1), N(3)–S(1)–N(4) 102.5(1), N(1)–S(2)–N(2) 105.1(1), N(1)–S(2)–N(5) 110.2(1), N(2)–S(2)–N(5) 102.4(1), N(2)–S(3)–N(3) 119.0(1), S(1)–N(1)–S(2) 111.8(1), S(2)–N(2)–S(3) 119.7(1), S(1)–N(3)–S(3) 121.5(1), S(1)–N(4)–C(1) 118.8(2), S(2)–N(5)–C(1) 120.8(2), N(4)–C(1)–N(5) 129.9(2), N(4)–C(1)–C(2) 115.6(2), N(5)–C(1)–C(2) 114.5.

$\text{PhCN}_2\text{S}_2+\text{Cl}^-$)⁴ the 1,3-dichloro-5-phenyl-1,3,2,4,6-dithiatriazine (2), which can be recrystallised from CH_2Cl_2 -hexane as yellow moisture-sensitive plates [2.52 g, 9.47 mmol, 60%, m.p. 119–121 °C; m/z (E.I., 70 eV) 230 ($\text{PhCN}_3\text{S}_2\text{Cl}^+$, 2%), 195 ($\text{PhCN}_3\text{S}_2^+$, 39), 181 ($\text{PhCN}_2\text{S}_2^+$, 11), 149 (PhCN_2S^+ , 46), 135 (PhCNS^+ , 13), 103 (PhCN^+ , 50), 46 (NS^+ , 100)]. Treatment of (2) with *N,N'*-bis(trimethylsilyl)sulphur diimide reforms the bicyclic structure (1). We note that the isolation of (2) completes the series of heterocycles $(\text{PhC})_x(\text{S}(\text{Cl}))_{3-x}\text{N}_3$ (for $x = 2$, see ref. 2).

Reduction of (2) (0.50 g, 1.87 mmol) with triphenylstibine (0.66 g, 1.87 mmol) in 50 ml chloroform yields a buff-coloured air-stable microcrystalline precipitate of 5-phenyl-1,3,2,4,6-thiatriazine (3) [0.28 g, 0.15 mmol, 78%, m.p. 115 °C; m/z (E.I., 70 eV) 195 ($\text{PhCN}_3\text{S}_2^+$, 19%), 181 ($\text{PhCN}_2\text{S}_2^+$, 6), 149 (PhCN_2S^+ , 21), 135 (PhCNS^+ , 3), 103 (PhCN^+ , 25), 92 (S_2N_2^+ , 12), 78 (S_2N^+ , 11), 76 (?), 64 (S_2^+ , 9), 46 (SN^+ , 100)]. We have confirmed the structural integrity of the six-membered CN_3S_2 ring in (3) by its reconversion into (2) (in 85% yield) upon treatment with chlorine and through formation of the norborna-1,3-diene adduct (4). The adduct is obtained (in 63% yield) by reacting a slurry of (3) (0.45 g, 2.3 mmol) in 10 ml acetonitrile with an excess of norbornadiene. The product can be recrystallized from acetonitrile as colourless plates [m.p. 179–180 °C; m/z (E.I., 70 eV) 287 ($\text{PhCN}_3\text{S}_2 \cdot \text{C}_7\text{H}_8^+$, 1%), 259 ($\text{PhCNS}_2 \cdot \text{C}_7\text{H}_8^+$, 2), 195 ($\text{PhCN}_3\text{S}_2^+$, 100), 149 (PhCN_2S^+ , 41), 103 (PhCN^+ , 21), 91 (C_7H_7^+ , 19); ¹H n.m.r. (400 MHz, CDCl_3) δ 7.3–7.9 (m, 5H, Ph), 6.24 (s, 2H, $\text{H}_{5,6}$), 4.48 (d, 2H, $\text{H}_{1,4}$, J 1.5 Hz), 3.24 (t, 2H, $\text{H}_{2,3}$, J 1.7 Hz), 2.08 and 1.16 (d, 1H, H_7 and H_8 , $J_{7,8}$ 9.8 Hz)].¹¹

The colour of (3) and its low solubility in organic media set it apart from the formally isoelectronic 8π -heterocycles $\text{R}_2\text{PN}_3\text{S}_2$ and $\text{S}_3\text{N}_3\text{O}_2^-$ that we have prepared.^{11,12} The CF_3 -substituted dithiatriazine $\text{CF}_3\text{CN}_3\text{S}_2$ has been reported recently, but no details were provided.^{5b} The behaviour of (3) leads us to suggest that the compound is not monomeric.

We thank the Natural Sciences and Engineering Research Council of Canada, the National Science Foundation, the Research Corporation and the State of Arkansas for financial support. R. T. B. acknowledges receipt of an N.S.E.R.C. postdoctoral fellowship.

Received, 9th April 1985; Com. 471

References

- H. Koenig and R. T. Oakley, *J. Chem. Soc., Chem. Commun.*, 1983, 73; P. W. Coddling, H. Koenig, and R. T. Oakley, *Can. J. Chem.*, 1983, **61**, 1562.
- A. W. Cordes, P. J. Hayes, P. D. Josephy, H. Koenig, R. T. Oakley, and W. T. Pennington, *J. Chem. Soc., Chem. Commun.*, 1984, 1021; P. J. Hayes, R. T. Oakley, A. W. Cordes, and W. T. Pennington, *J. Am. Chem. Soc.*, 1985, **107**, 1346.
- S. T. A. K. Daley, C. W. Rees, and D. J. Williams, *J. Chem. Soc., Chem. Commun.*, 1984, 55, 57.
- A. J. Banister, N. R. M. Smith, and R. G. Hey, *J. Chem. Soc., Perkin Trans. 1*, 1983, 1181.
- (a) H-U. Hofs, R. Mews, W. Clegg, M. Noltemeyer, M. Schmidt, and G. M. Sheldrick, *Chem. Ber.*, 1983, **110**, 416; (b) H-U. Hofs, G. Hartmann, R. Mews, and G. M. Sheldrick, *Z. Naturforsch., Teil B*, 1984, **39**, 1389.
- G. K. Maclean, J. Passmore, M. J. Schriver, P. S. White, D. Bethell, R. S. Pilkington, and L. H. Sutcliffe, *J. Chem. Soc., Chem. Commun.*, 1983, 807.
- H. W. Roesky, P. Schafer, M. Noltemeyer, and G. M. Sheldrick, *Z. Naturforsch., Teil B*, 1983, **38**, 347.
- I. Ernest, W. Hollick, G. Rihs, D. Schomburg, G. Shoham, D. Wenkert, and R. B. Woodward, *J. Am. Chem. Soc.*, 1981, **103**, 1540.
- A. R. Sanger, *Inorg. Nucl. Chem. Lett.*, 1973, **9**, 351.
- J. Weiss, I. Ruppert, and R. Appel, *Z. Anorg. Allg. Chem.*, 1974, **406**, 329.
- N. Burford, T. Chivers, A. W. Cordes, W. G. Laidlaw, M. C. Noble, R. T. Oakley, and P. N. Swepston, *J. Am. Chem. Soc.*, 1982, **106**, 1282.
- T. Chivers, A. W. Cordes, R. T. Oakley, and W. T. Pennington, *Inorg. Chem.*, 1983, **22**, 2429.